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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.	
09/990,049 11/21/2001		William Ford	450117-03449	1484	
20999	7590 01/23/2006		EXAMINER		
FROMMER LAWRENCE & HAUG 745 FIFTH AVENUE- 10TH FL.			NAFF, DAVID M		
NEW YORK, NY 10151			ART UNIT	PAPER NUMBER	
	•		1651		

DATE MAILED: 01/23/2006

Please find below and/or attached an Office communication concerning this application or proceeding.

		Application	ı No.	Applicant(s)		
		09/990,049		FORD ET AL.		
Office Action Summary		Examiner		Art Unit		
		David M. Na	aff	1651		
The MAILI Period for Reply	NG DATE of this communication	appears on the o	over sheet with the c	correspondence a	ddress	
A SHORTENED S WHICHEVER IS Extensions of time mater SIX (6) MONTHS If NO period for reply Failure to reply within Any reply received by	STATUTORY PERIOD FOR RELONGER, FROM THE MAILING by be available under the provisions of 37 CFF from the mailing date of this communication is specified above, the maximum statutory bethe set or extended period for reply will, by stitle office later than three months after the migustment. See 37 CFR 1.704(b).	G DATE OF THIS R 1.136(a). In no event b. riod will apply and will of atute, cause the applica	S COMMUNICATION t, however, may a reply be tin expire SIX (6) MONTHS from ation to become ABANDONE	N. nely filed the mailing date of this o D (35 U.S.C. § 133).		
Status						
2a)⊠ This action 3)□ Since this a	e to communication(s) filed on <u>1</u> is FINAL . 2b) 1 application is in condition for allocation to the practice und	This action is not wance except for	n-final. or formal matters, pro		e merits is	
Disposition of Clain	ns					
4a) Of the a 5) ☐ Claim(s) 6) ☑ Claim(s) 25 7) ☐ Claim(s)	i-45 and 47 is/are pending in the bove claim(s) is/are with is/are allowed. is/are allowed. is-45 and 47 is/are rejected. is/are objected to. are subject to restriction ar	drawn from cons				
Application Papers						
9)☐ The specific 10)☑ The drawing Applicant ma Replacemen	ration is objected to by the Example; (s) filed on 21 November 2001 by not request that any objection to the drawing sheet (s) including the condeclaration is objected to by the	is/are: a) acc the drawing(s) be rrection is required	held in abeyance. Sed if the drawing(s) is ob	e 37 CFR 1.85(a). jected to. See 37 C	CFR 1.121(d).	
Priority under 35 U.	S.C. § 119					
 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) All b) Some * c) None of: 1. Certified copies of the priority documents have been received. 2. Certified copies of the priority documents have been received in Application No 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received. 						
Attachment(s)						
	on's Patent Drawing Review (PTO-948) ure Statement(s) (PTO-1449 or PTO/SB	3/08)	4) Interview Summary Paper No(s)/Mail D 5) Notice of Informal F 6) Other:	ate	⁻ O-152)	

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DETAILED ACTION

An amendment of 10/14/05 to an office action of 4/14/05 amended claims 25, 26, 29, 33, 35, 37, 38, 40 and 41, and canceled claim 46.

Claims examined on the merits are 25-45 and 47, which are all claims in the application.

The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action.

Claim Rejections - 35 USC § 112

Claims 25-39 and 43-45 are rejected under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention.

The claims are confusing and unclear by claim 25 (line 13) being unclear as to the relationship of the "interactive group" to reacting required in line 12. The substance that contains the interactive group and the function of the interactive group during reacting is unclear. In line 13, the portion of the line from "or" to part of "through" has been lined through. Is this intended?

In line 2 of claim 38, "are" should be changed to --- is --- since "nanoparticle of the composite" is recited.

Response to Arguments

Applicants point out that amendments render the 35 U.S.C. 112, second paragraph rejection moot. However, indefiniteness still remains as pointed out above.

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Claim Rejections - 35 USC § 103

Claims 25-31, 33-45 and 47 are rejected under 35 U.S.C. 103(a) as being unpatentable over Pompe et al (AR) in view of Singh et al (5,560,960) and Richter et al (AQ) for reasons in the previous office action of 4/14/05, and for reasons herein.

The claims are drawn to a process of producing a metal nanoparticle-nucleic composite by reacting a nucleic acid with a metal complex to produce a metal complex-nucleic acid conjugate by metalation of bases and/or binding through an interactive group, removing non-conjugated byproducts, and reacting the conjugate with a reducing agent to produce the metal nanoparticle-nucleic acid composite. The metal nanoparticle of the composite is catalytically active towards electroless metallisation. Also claimed is a metal nanoparticle-nucleic acid composite resulting from the process, a process of making a nanowire by treating the composite by electroless deposition of metal, a nanowire resulting from the process, and a network or electronic circuit containing the nanowire.

Pompe et al disclose (page 1090, left col, second full paragraph) that Pt(II) and Pd(II) complexes such as cis-diamminedichloroplatinum attach to DNA bases to form stable monofunctional and bifunctional adducts. Further disclosed (third full paragraph of the left col) is that the Pt-DNA bond is not broken during reduction, and that Pt(II) and Pd(II) complexes attached to DNA double chain can act as nucleation centers for the growth of metal clusters. Also disclosed is carrying out metallization of DNA by adding DNA to Pd salt solution

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followed by adding a reducing agent, and obtaining clusters on the DNA of 3 to 5 nm in diameter in a few seconds after adding the reducing agent (paragraph bridging the cols, page 1090).

Singh et al disclose (paragraph bridging cols 1 and 2)

precipitating nanometer-sized metal particles from solution within

vesicles made from polymerizable phospholipids. Polymerized

phospholipids are formed and added to a electroless plating solution.

Before the electroless plating solution is added, palladium or

platinum is provided on the inside surface of vesicles to function as

a catalyst (col 3, lines 44-64). To insure that metal particles form

only on the inside surface, any metal on the exterior surface of the

vesicle is removed such as by using a chelating agent and gel

filtering, or by passing the vesicles through an ion exchange column.

Singh et al further disclose (col 5, line 18) using cobalt, nickel or

iron when producing metal nanoparticles by electroless plating.

Richter et al disclose (page 508 and 510) metallization of DNA similar to Pompe et al and disclose formation of clusters of 1-5 nm diameter on DNA (page 508, left col, third full paragraph).

It would have been obvious to attach cis-diamminedichloroplatinum to DNA as disclosed by Pompe et al, and then use a reducing agent to obtain DNA containing attached platinum metal catalysis for use in electroless deposition of metal on the DNA as suggested by Singh et al subjecting vesicles containing Pd or Pt to electroless metal deposition and as suggested by Pompe et al carrying out metallization of DNA by treating a DNA solution with a Pd salt solution, and then

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adding a reducing agent to form metal clusters on the DNA. Removing any non-attached metal complex from the DNA before electroless metallization would have been obvious to prevent the non-attached metal complex from forming metal particles as suggested by Singh et al removing metal from the exterior of vesicles to prevent metal particles from being formed on the vesicles exterior surface. objective of Pompe et al is to obtain metal clusters on the DNA and not at other places, and to accomplish this one would obviously have to remove non- attached metal complex before electroless metallization. Removing any non-conjugated by-products would have 10 been obvious simply to prevent any possible inference with subsequent It would have been apparent form Richter et al that metal reactions. clusters of 1-5 nm diameter can be obtained, and it would have been obvious to produce clusters not thicker than DNA since this is an objective of Pompe et al (page 1090, left col, first full paragraph). 15 Reacting DNA with cis-diamminedichloroplatinum as disclosed by Pompe et al followed by reducing as set forth above will inherently result in metallization of bases, and provide a metal nanoparticle active towards electroless metallization. When carrying out metallization of DNA as set forth above, it would have been obvious to form a nanowire 20 since Pompe et al (page 1090, right col, lines 1-10) and Richter et al (paragraph bridging pages 508 and 509) obtain a nanowire. Using the nanowire in an electronic circuit would have been obvious since metal wires are conventionally used in such circuits. The metallization of Pompe et al and Richter et al is controlled since they disclose

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controlling the time of metallization to control the size of clusters. The use of cobalt, nickel or iron when producing metal nanoparticles by electroless plating as disclosed by Singh et al would have suggested using a electroless plating solution as in claim 38.

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Response to Arguments

Applicants urge that the combination of references do not suggest the feature of claim 25 that recites "wherein the metal complex-nucleic acid conjugate is formed by the specific reacting of the nucleic acid specific metal complex with bases of the nucleic acid".

However, in Pompe et al and Richter et al, reacting a metal complex with DNA, inherently results in the metal complex reacting with bases of the nucleic acid.

Claim Rejections - 35 USC § 103

Claim 32 is rejected under 35 U.S.C. 103(a) as being unpatentable over the references as applied to claims 25-31, 33-45 and 47 above, and further in view of Newsman et al (5,670,680) for reasons in the previous office action, and for reasons herein.

The claim requires a gaseous reducing agent.

Singh et al disclose using hydrogenation (col 4, line 57) for reducing metal ions to produce metals in a process of producing metal nanoparticles by electroless plating.

Newman et al disclose using hydrogen gas for hydrogenation in producing metal complexes.

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It would have been obvious to use hydrogen gas as a reducing agent to reduce the metal of a conjugate of a metal complex and DNA disclosed by Pompe et al as suggested by Singh et al and Newman et al.

Response to Amendment

This rejection has not been separately traversed, other than rely on arguments presented in response to the rejection above.

Double Patenting

Claims 25-45 and 47 are provisionally rejected under the judicially created doctrine of obviousness-type double patenting as being unpatentable over claims 1-4, 6-22 and 25-35 of copending Application No. 10/210,812 in view of Singh et al.

The claims of the copending application require metallization of a nucleic acid to produce a metal nanoparticle-nucleic acid composite.

It would have been obvious in view of Singh et al for the type of reasons set forth above to remove non-conjugated metal complexes and/or non-conjugated by-products, if formed, before treatment with a reducing agent in the process of the copending application claims for metallization of DNA. The presence of extraneous metal complex or other by-products will obviously be a contaminant that can interfere with subsequent reactions.

This is a <u>provisional</u> obviousness-type double patenting rejection because the conflicting claims have not in fact been patented.

Response to Arguments

Applicants' urge that the copending application claims do not remove non-conjugated metal complex, and there is not motivation in

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the '812 application claims to combine its teachings with Singh et al. However, the claims recite "removing any", which requires removing non-conjugated metal complexes and/or non-conjugated products only if formed. The claims do not require the non-conjugated complexes or products to be formed. Furthermore, removal of non-conjugated metal complex that is not at a desired site for depositing metal is suggested by Singh et al, and the reason for removing extraneous metal complex provided by Singh et al is motivation.

Conclusion

Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, THIS ACTION IS MADE

FINAL. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to David M. Naff

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whose telephone number is 571-272-0920. The examiner can normally be reached on Monday-Friday 9:30-6:00.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Mike Wityshyn can be reached on 571-272-0926. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

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